ABSTRACT

Objective: This study aimed to evaluate the suitability of one specific silorane-based composite for the application as a repair material for different substrates.

Materials and methods: Truncated cones of composites fabricated and thermocycled for 6000 cycles to serve as a substrate were made of two commercially available dimethacrylate-based composites (DBC) (Filtek Supreme and Clearfil APX) and one silorane-based composite (SBC) (Filtek P90). The surface of the specimens was treated with air-abraded using Al₂O₃-particles of 50 μm size. A Teflon device was used to fabricate inverted truncated cones of repair composite over the surface-treated top of each original truncated cone. DBC and SBC were used as repair materials and bonded to the specimens using either a dimethacrylate-based (single bond 2) or a phosphate-methacrylate-based (adhesive belongs to the silorane) adhesives. The specimens were stored in distilled water at 37ºC for 7 days and stressed to failure under tension. The data were analyzed with one-way ANOVA and Tukey tests at 5%.

Results: Bonding DBC as a repair material to a DBC substrate using the dimethacrylate-based adhesive produced the highest bond strength. Bonding DBC as repair material using the phosphate-methacrylate-based adhesive or SBC as repair material using the dimethacrylate-based adhesive produced lower bond strength, regardless the brand and of the chemical formulation of the aged substrate.

Conclusion: In order to obtain high bond strength, there is need to match adhesive and repair material, regardless the brand and of the chemical formulation of the aged substrate.

Keywords: Composite, Silorane, Adhesion, Bond strength.

INTRODUCTION

Failure or fracture defects of composite restorations should be first evaluated for the possibility of repair procedure instead of complete removal and remake. Combinations of surface treatments and bonding agents influence the bond strength of the repaired dimethacrylate-based composites. Junior et al (2008) observed that the use of aluminium oxide sandblasting on the aged composite surface enhances the shear-bond strength of repaired dimethacrylate-based composites. Application of a thin layer of bonding agent on the aged composite surface increased significantly the tensile strength of the repair procedure. Also, the roughening of filling surfaces associated to application of a low-viscosity resin enhance the repair bond strength.

Padipatvuthikul (2007) suggests that the mechanisms of adhesion of repair procedure promoted by resin-based bonding agents can be due to the micromechanical retention through monomer penetration into surface irregularities; and due to the chemical bond formation promoted by solvents in the adhesive systems. The solvents can cause swelling and gelation of the surface layer, allowing the monomer penetrate to the unconverted vinyl groups (—C= = C) in the aged composite subsurface. However, most authors ‘repaired’ the original composite with composites with same chemical formulation (dimethacrylate-based).

Recently, silorane-based composite (SBC) (Filtek Silorane, 3M ESPE Dental Products) was introduced in the dental market. The monomer system consists a functional epoxy groups that contain both siloxane and oxirane structural moieties (i.e. ‘silorane’) using ring opening polymerization of the silorane molecules. This composite was designed to reduce shrinkage stress and cuspal deflection associated with conventional dimethacrylate-based composite (DBC).

It should be important to note the compatibility of composites with different chemical formulation, when repair procedure is indicated. Also, it is necessary to define a repair protocol between DBCs and SBCs. The aim of this study was to evaluate the suitability of one specific silorane-based composite for the application as a repair material for different substrates.
MATERIALS AND METHODS

The materials used in this study are listed in Table 1.

A two-piece Teflon device was used to prepare truncated cones of composites fabricated to serve as a substrate of two commercially available DBCs (Filtek Supreme and Clearfil APX, shade A3) and one SBC (Filtek P90, shade A3). Each truncated cone measured 4 mm in height, with the bottom being 4 mm in diameter and the top being 2 mm in diameter. The composites were inserted in the Teflon device in 2 mm thick increments. Each composite increment was polymerized for 20 seconds with a light curing unit (XL 3000, 3M ESPE) at an output intensity of 500 mW/cm².

One hundred and thirty-five truncated resin composite cones were prepared and stored in distilled water at 37°C for 7 days. The truncated cones were thermal-cycled for 6,000 cycles at a temperature ranging between 5°C and 55°C (± 2°C), with a dwell time of 30 seconds, followed by storage in distilled water at 37°C for another 24 hours.

The surface of the piece of composites was air-abraded using Al₂O₃-particles of 50 μm size according to the manufacturer’s recommendations (Microetcher ERC, Danville Engineering, California, USA). DBC and SBC were used as repair materials and bonded to the surface of the piece of composites air-abraded using either a dimethacrylate-based (single bond 2 adhesive system, 3M ESPE) or a phosphate-methacrylate-based (silorane system adhesive bond, 3M ESPE—adhesive belongs to the silorane) adhesives.

A Teflon device was used to fabricate an inverted truncated cone of repair composite over the surface-treated top of each original aged truncated cone, using incremental applications, with each layer polymerized separately for 20 seconds. The final specimen consisted of two inverted cones.

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Table 1: The materials used in this study

<table>
<thead>
<tr>
<th>Material</th>
<th>Manufacturer</th>
<th>Composition</th>
</tr>
</thead>
<tbody>
<tr>
<td>Adper single bond 2 adhesive system</td>
<td>3M ESPE, St Paul, MN, USA</td>
<td>Bis-GMA (bisphenol-a-glycidyl dimethacrylate), HEMA (2-hydroxyethyl methacrylate), dimethacrylate, methylacrylate functional copolymer of polyacrylic and polytaconic acid, water, alcohol, photoinitiator</td>
</tr>
<tr>
<td>Silorane system adhesive bond</td>
<td>3M ESPE, St Paul, MN, USA</td>
<td>TEGDMA (triethylene glycol dimethacrylate), phosphoric acid methacryloxy-hexylesters, 16-hexanediol dimethacrylate</td>
</tr>
<tr>
<td>Filtek Supreme</td>
<td>3M ESPE, St Paul, MN, USA</td>
<td>BIS-GMA, BIS-EMA (Bisphenol A polyethylene glycol diether dimethacrylate), UDMA (urethane dimethacrylate), TEGDMA. Nonagglomerated/non-aggregated 75 nm silica nanofiller, agglomerate silica nanocluster silica nanoparticles of 75 nm</td>
</tr>
<tr>
<td>Filtek P90</td>
<td>3M ESPE, St Paul, MN, USA</td>
<td>3,4-epoxycyclohexylethylcyclopolymermethylsiloxane (5-15 wt%), bis-3,4-epoxycyclohexylethyl-phenylmethylsilane (5-15 wt%), silanized quartz (50-70 wt%), yttrium fluoride (10-20 wt%)</td>
</tr>
<tr>
<td>Clearfil APX</td>
<td>Kuraray Co., Ltd, Japan</td>
<td>Bis-GMA, TEGDMA, silanated barium glass filler, silanated silica filler, silanated colloidal silica, dl-camphorquinone, catalysts, accelerators, pigments</td>
</tr>
</tbody>
</table>
truncated cones of resin composites united by their circular top surfaces, where the repair was made (Fig. 1A).10

Fifteen specimens for each group were prepared (Table 2). The specimens were stored in distilled water at 37ºC for 7 days.2

The specimens were stress to failure (Fig. 1B) under tension using a universal testing machine (Emic DL 2000, São José dos Pinhais, PR, Brazil) with a load cell of 50 kgf at 1 mm/min.

After the tensile test, the specimens were analyzed with a 20× stereomicroscope (Stemi 2000, Karl Zeiss, Germany). Fractures were classified as cohesive in the substrate, adhesive at the interface or cohesive in the adherend.

Repair bond strengths (in MPa) derived from the nine groups were first examined to evaluate the normality (Shapiro-Wilk test) and homoscedasticity (modified Levene test) of the acquired data. As those assumptions did not appear to have been violated, the data was further analyzed using one-way ANOVA and Tukey multiple comparison tests. Statistical significance for all tests were preset at $\alpha = 0.05$.

### RESULTS

A highly significant difference ($p = 0.0000$; $F = 288$; $DF = 8$) was detected among the repair bond strengths of the nine groups. Pairwise comparisons of the repair bond strengths are shown in Table 3. Group 1 (control dimethacrylate—DBC substrate and dimethacrylate-based adhesive and DBC restoration) exhibited the highest repair strength which was not significantly different ($p > 0.05$) from the groups 4 and 5 (DCB substrate plus phosphate-methacrylate-based adhesive and SBC restoration) and Group 2 (control silorane—SBC substrate and phosphate-methacrylate-based adhesive and SBC restoration). The repair strength of groups 8 and 9 (DBC substrate and dimethacrylate-based adhesive and SBC restoration), Groups 7 and 6 (SBC substrate and phosphate-methacrylate-based adhesive and DBC restoration) and group 3 (SBC substrate and dimethacrylate-based adhesive and SBC restoration) was lower statistically different from those derived from groups 1, 2, 4 and 5.

### Table 2: Experimental groups evaluated in this study

<table>
<thead>
<tr>
<th>Group</th>
<th>Substrate</th>
<th>Intermediate layer (adhesive)</th>
<th>Composite repair</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Dimethacrylate (Filtek Supreme)</td>
<td>Dimethacrylate-based</td>
<td>Dimethacrylate (Filtek Supreme)</td>
</tr>
<tr>
<td>2</td>
<td>Silorane (Filtek P90)</td>
<td>Phosphate-methacrylate-based</td>
<td>Silorane (Filtek P90)</td>
</tr>
<tr>
<td>3</td>
<td>Silorane (Filtek P90)</td>
<td>Dimethacrylate-based</td>
<td>Silorane (Filtek P90)</td>
</tr>
<tr>
<td>4</td>
<td>Dimethacrylate (ClearFil APX)</td>
<td>Phosphate-methacrylate-based</td>
<td>Silorane (Filtek P90)</td>
</tr>
<tr>
<td>5</td>
<td>Silorane (Filtek P90)</td>
<td>Dimethacrylate-based</td>
<td>Silorane (Filtek P90)</td>
</tr>
<tr>
<td>6</td>
<td>Dimethacrylate (Filtek Supreme)</td>
<td>Dimethacrylate-based</td>
<td>Dimethacrylate (Filtek Supreme)</td>
</tr>
<tr>
<td>7</td>
<td>Silorane (Filtek P90)</td>
<td>Phosphate-methacrylate-based</td>
<td>Silorane (Filtek P90)</td>
</tr>
<tr>
<td>8</td>
<td>Dimethacrylate (Filtek Supreme)</td>
<td>Dimethacrylate-based</td>
<td>Silorane (Filtek P90)</td>
</tr>
<tr>
<td>9</td>
<td>Silorane (Filtek P90)</td>
<td>Phosphate-methacrylate-based</td>
<td>Dimethacrylate (ClearFil APX)</td>
</tr>
</tbody>
</table>

### Table 3: Tensile bond strength (MPa) means and standard deviations of the groups

<table>
<thead>
<tr>
<th>Groups</th>
<th>Substrate</th>
<th>Intermediate layer (adhesive)</th>
<th>Composite repair</th>
<th>Mean in MPa (±SD)</th>
<th>Homogeneous sets*</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Dimethacrylate (Filtek Supreme)</td>
<td>Dimethacrylate-based</td>
<td>Dimethacrylate (Filtek Supreme)</td>
<td>20.98 (±2.24)</td>
<td>A</td>
</tr>
<tr>
<td>5</td>
<td>Dimethacrylate (Filtek Supreme)</td>
<td>Phosphate-methacrylate-based</td>
<td>Silorane (Filtek P90)</td>
<td>19.71 (±2.17)</td>
<td>A</td>
</tr>
<tr>
<td>4</td>
<td>Dimethacrylate (ClearFil APX)</td>
<td>Phosphate-methacrylate-based</td>
<td>Silorane (Filtek P90)</td>
<td>19.51 (±1.61)</td>
<td>A</td>
</tr>
<tr>
<td>2</td>
<td>Silorane (Filtek P90)</td>
<td>Phosphate-methacrylate-based</td>
<td>Silorane (Filtek P90)</td>
<td>19.45 (±2.38)</td>
<td>A</td>
</tr>
<tr>
<td>9</td>
<td>Dimethacrylate (ClearFil APX)</td>
<td>Dimethacrylate-based</td>
<td>Silorane (Filtek P90)</td>
<td>4.44 (±2.19)</td>
<td>B</td>
</tr>
<tr>
<td>6</td>
<td>Silorane (Filtek P90)</td>
<td>Phosphate-methacrylate-based</td>
<td>Dimethacrylate (Filtek Supreme)</td>
<td>4.41 (±1.02)</td>
<td>B</td>
</tr>
<tr>
<td>8</td>
<td>Dimethacrylate (Filtek Supreme)</td>
<td>Dimethacrylate-based</td>
<td>Silorane (Filtek P90)</td>
<td>3.88 (±1.67)</td>
<td>B</td>
</tr>
<tr>
<td>3</td>
<td>Silorane (Filtek P90)</td>
<td>Dimethacrylate-based</td>
<td>Silorane (Filtek P90)</td>
<td>3.54 (±1.46)</td>
<td>B</td>
</tr>
<tr>
<td>7</td>
<td>Silorane (Filtek P90)</td>
<td>Phosphate-methacrylate-based</td>
<td>Dimethacrylate (ClearFil APX)</td>
<td>3.53 (±1.43)</td>
<td>B</td>
</tr>
</tbody>
</table>

SD: Standard of deviation; *The groups with the same letters show no significant differences
Examination of specimens after fracture indicated predominantly adhesive fractures for all groups (>95%).

DISCUSSION

Two composites (Clearfil APX, Filtek Supreme) selected for repair (‘substrate’), have dimethacrylate resin matrices (bis-GMA) but in differing amounts. One non-dimethacrylate hybrid material (Filtek P90 silorane composite) was selected, as the resin matrix is a ring-opening monomer. SBCs were developed for reduce problems of polymerization shrinkage, polymerization stress and water sorption. However, in case of the possibility of repair, a recent dilemma for the clinician is that composite monomers other than dimethacrylates (e.g. siloranes) are being used which require compatibility of materials differing in composition of matrix. Therefore, this study investigated the ‘repairability’ of composite restorations with different chemical formulation.

There is no consensus for the aging regimens simulating the oral conditions. In this study, 30 days short-term water storage followed by 6000 thermal cycles, then 24 hours water storage were used to obtain an aged substrate surface. The aging of the substrates has the purpose to simulate the clinical service changes that occurs in the filling, such as a decrease of unreacted methacrylate groups, water sorption and leaching of components. According to Maneenut et al (2010), Silorane system adhesive bond is not silorane-based but phosphate-dimethacrylate-based. Confirming results of this study, Tezvergil-Mutluay et al (2008) found an increase in shear bond strength between SBC and DBC using a phosphate-dimethacrylate-based intermediate resin compared to a dimethacrylate-based intermediate layer. Also, Moser et al (2009) observed that the use of a phosphate-dimethacrylate-based intermediate layer increases the shear bond strength of repairs between SBCs and DBCs.

However, Maneenut et al (2010) found that SBC repair material associated with the silane-based adhesive (Clearfil repair) showed highest bond strength then silorane system adhesive bond was used as an intermediate layer. According to Ivanovas et al (2010), mixed repairs do not produce lower bond strengths than repairs made of one sort of composite resin, SBC or DBC. They found that the mixed repairs with SBC and DBC using a silane coupling agent seem to be more reliable.

As emphasized by the manufacturers of Filtek P90 composite and Silorane system adhesive bond, their silorane system is chemically incompatible with dimethacrylate systems. Confirming the results of this study, Tezvergil-Mutluay et al (2008) found that the bond strength between SBC and DBC without any intermediate resin showed lowest values compared silorane-silorane and dimethacrylate-dimethacrylate combinations without intermediate layer.

Fracture pattern of specimens on all groups showed that there were predominantly adhesive fracture patterns along the aged substrate/intermediate layer/fresh composite. This result can be attributed to the microtensile test, suggested by Sano et al. They assess the tensile bond strength of specimens with small surface areas of the adhesive joint, where bond failures essentially occur at the adhesive interface. This method permits measurement of bond strengths of different test methods without cohesive failure. If the clinician is unable to identify the present filling material, appropriate repair technique should be utilized.
It is important to emphasize that a SBC repair material with a silorane-based adhesive intermediate layer should be used, and conversely DBC repair material with a dimethacrylate-based intermediate layer should be used to provide a satisfactory bond in repair procedures.

CONCLUSION

In order to bonding DBC as a repair material to a SBC substrate, a phosphate-methacrylate-based intermediate layer is required. In order to bonding SBC as a repair material to a DBC substrate, a dimethacrylate-based adhesive layer is required.

REFERENCES


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